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# Low-carbon Methanol Value Chain

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### **Glossary**

atm Atmospheres

bara Bar absolute pressure
Bcm Billion cubic meters
BFD Block flow diagram
BLI Battery limits investment
Btu British thermal units

¢/MMBtu Cents per million British thermal units

¢/gal Cents per gallon¢/kWh Cents per kilowatt-hour¢/lb Cents per pound

¢/Mgal Cents per thousand gallons¢/Mlb Cents per thousand pounds

¢/Mscf Cents per thousand standard cubic feet

¢/scf Cents per standard cubic footCCS Carbon capture and storage

cP Centipoise

CR Controlled rheology

cf/ft<sup>3</sup> Cubic feet

\$/t Dollars per metric ton
DAC Direct air capture
DME Dimethyl ether

g Grams gal Gallons

gpm Gallons per minute

h Hour(s)
kg Kilograms
kJ Kilojoules
KO Knockout
kPa Kilopascals
kWh Kilowatt-hours
lb Pounds

Mgal Thousand gallons
Mlb Thousand pounds
MMlb/y Million pounds per year
MMt Million metric tons

MMt/y Million metric tons per year

mol% Molar percent
mPa Megapascals
MTG Methanol-to-gasoline
MTO Methanol-to-Olefins
MTP Methanol-to-Propylene
OSI Off-sites investment

PEP Process Economics Program

PFD Process flow diagram
PP Polypropylene
ppb Parts per billion
ppm Parts per million

psi Pounds per square inch

psia Pounds per square inch absolute psig Pounds per square inch gauge

ROI Return on investment RWGS Reverse water gas shift

s Second(s)

scf Standard cubic feet scm Standard cubic meters syngas Synthesis gas t Metric ton

t/y Metric tons per year TEAI Triethyl aluminum TFC Total fixed capital

VOC Volatile organic compound

vol% Volume percent wt% Weight percent w/w Weight for weight

y Year

## **Abstract**

Methanol is a crucial feedstock in industrial chemistry, playing an important role in the synthesis of various chemicals, such as formaldehyde (HCHO), acetic acid (CH<sub>3</sub>COOH), and methyl methacrylate (C<sub>5</sub>H<sub>8</sub>O<sub>2</sub>). These chemicals are essential for producing polymers, plastics, and advanced materials. In recent years, particularly since the early 2010s, interest in low-carbon methanol has significantly increased. This trend gained considerable momentum from 2018 onward due to substantial investments and technological innovations in the low-carbon methanol industry. The growing awareness of climate change and urgent need for sustainable energy alternatives have highlighted the importance of decarbonization and transitioning to renewable energy sources.

Supportive policies and regulations that encourage the adoption of low-carbon fuels have further accelerated the development and acceptance of low-carbon methanol. This aligns with global initiatives aimed at achieving net-zero carbon emissions, particularly in industries like transportation and chemicals. Consequently, these factors have collectively increased interest in low-carbon methanol as a viable alternative to conventional fossil fuels.

This review analyzes the production economics of three sets of integrated processes, starting with biomethanol as a raw material. The integrated processes include the conversion of biomethanol to acetic acid and then to acetic anhydride, biomethanol to formaldehyde and then to resol syrup, and biomethanol to propylene and then to polypropylene, with production capacities of 750 million pounds per year (lb/y) for acetic anhydride, 80 million lb/y for resol syrup, and 1,199 million lb/y for polypropylene, respectively. The production economics discussed are based on a US Gulf Coast location and data from the first quarter of 2025, using a PEP Cost Index of 1,634. Additionally, this review will assess the carbon and water footprints of the integrated processes.

The technoeconomic assessments of these processes represent PEP's independent interpretation of commercial processes, relying on data from publicly available sources such as patents and technical journals. We believe that our analysis and process economics provide a sufficiently accurate representation for a conceptual process design assessment, although they may not completely reflect the actual plant configuration.

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